CORONOPILIN—ANOTHER MAJOR SESQUITERPENE LACTONE IN PARTHENIUM HYSTEROPHORUS

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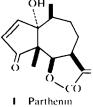
INTRODUCTION

Parthenium hysterophorus L. (Compositae), an aggressive weed of the Americas including the Carribbean, has spread in the last hundred years to Australia, Africa and Asia [1]. In some parts of India it is responsible for a high incidence of allergic contact dermatitis [2]. Parthenin (1) was isolated and identified as the major sesquiterpene lactone of this species [3-5] and shown to be the allergen responsible [2]. More recently, other sesquiterpene lactones, tetraneurin-A and the newly described hysterophorin ([6] the structures shown in this reference are incorrect) have been identified as additional allergens from P. hysterophorus growing in India. Using a new TLC method for visualization of sesquiterpene lactones [7], we have identified coronopilin (2) as the second major sesquiterpene lactone (after parthenin) and tetraneurin-A as a minor sesquiterpene lactone in samples of the plant from both India and Belize, central America.

RESULTS AND DISCUSSION

Coronopilin (identified by NMR, mp, R_f and colour on TLC) was found in extracts of dried stems, leaves, flowering heads, trichomes (Belize material) and fresh parts of P. hysterophorus grown at UBC. It was also isolated from achenes (Belize) as well as trichome and pollen extracts of Indian samples. The ratio of parthenin to coronopilin is approximately 10:1 (NMR). Coronopilin has been reported from numerous species of Ambrosia and Parthenium [8, 9] but not P. hysterophorus probably because it cannot be easily distinguished from parthenin by conventional methods of TLC analyses for sesquiterpene lactones (i.e. UV light, iodine vapours, or aqueous KMnO₄). However, more recently, coronopilin has also been identified in this species by Indian workers (A. Lonkar, personal communication).

The fact that coronopilin occurs in *P. hysterophorus* in appreciable amounts and that it is a strong allergen [10]



2 Coronopilin

might explain the reported stronger positive test reaction of patients to acetone extracts of this plant than to parthenin alone [11].

In addition to parthenin and coronopilin, small quantities of tetraneurin-A also occur in shoot and flower extracts but not in the pollen of *P. hysterophorus*. We did not detect ambrosin in any part in spite of its reported occurrence in trichomes [12]. Plant samples from India and Belize are very similar in their sesquiterpene lactone composition but material from various regions in South America (e.g. Argentina) vary considerably in chemical composition [13]. This, therefore, does not support the suggestion that *P. hysterophorus* in India originated from Argentina [14].

EXPERIMENTAL

Isolation and identification of coronopilin. Parthenin, isolated from shoots of dried P. hysterophorus (collected in Belize) according to the procedure of Rodriguez [15], was found on TLC (Polygram Si gel; CHCl₃-Me₂CO, (6:1); vanillin spray reagent [7]) to be a mixture of two compounds. 60 mg of the mixture (in CHCl₃) was chromatographed on a Si gel column and eluted with CHCl₃ Me₂CO (6:1). Early fractions gave a spot on TLC $(R_1 = 0.46)$ which reacted with the vanillin reagent to give a blue colour. The combined fractions yielded fine white crystals, mp 177°, ¹H NMR (CDCl₃ with TMS): δ 6.29 (d, H-13b), 5.61 (d, H-13a), 4.95 (d, H-6), 1.17 (s, C-10-Me), 1.10 (s, C-5-Me). The NMR and mp correspond to the reported values of coronopilin [16] and R_{ℓ} and colour on TLC were identical with those of an authentic sample of coronopilin (a gift from Prof. T. A. Geissman). Later fractions ($R_f = 0.44$, bluish-green colour with vanillin reagent) yielded pure parthenin, mp 167°, ¹H NMR(CDCl₃ with TMS): δ 7.50 (d, H-2), 6.25 (d, H-3), 6.33 (d, H-13b), 5.62 (d, H-13a), 5.02 (d, H-6), 1.30 (s, C-5-Me), 1.14 (s, C-10-Me). The parthenincoronopilin mixture before chromatography had mp 164° and NMR (CDCl₃ with TMS) identical with that given for parthenin

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ACYCLIC DITERPENES FROM CROTON KERRII

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Key Word Index—Croton kerrii; Euphorbiaceae; acyclic diterpenes; anti-reserpine ulcer; (E, E, Z)-11-hydroxymethyl-3,7,15-trimethyl-2,6,10,14-hexadecatetraen-1-ol; (E, E, E)-11-formyl-3,7,15-trimethyl-2,6,10,14-hexadecatetraen-1-ol.

INTRODUCTION

Recently, we reported the structures of 18-hydroxygeranylgeraniol (1) [1], an anti-reserpine ulcer substance, and of plaunols A,B,C,D and E [2,3], anti-Shay ulcer compounds, isolated from a Thai medicinal plant, Croton sublyratus Kurz. We have now commenced a chemotaxonomical survey of plants of the genus Croton which grow wild in Thailand. From Croton sp., various diterpenes belonging to the labdane (or clerodane) [4–10], pimarane [11], tiglane [12], crotofolane [13], and acyclic diterpene [1] have been isolated. We now report the isolation and characterization of two novel acyclic diterpenes from Croton kerrii A. Shaw (Euphorbiaceae) [14].

RESULTS AND DISCUSSION

A methanol extract of the leaves of $C.\,kerrii$ was washed with n-hexane. The n-hexane extract was subjected to silica gel chromatography to yield diterpenes 2a and 3a. The diterpene diol (2a) had a molecular formula of $C_{20}H_{34}O_2$ by high resolution MS of its bis-trimethylsilylether (Calc. for $C_{26}H_{50}O_2Si_2$: 450, 3349. Found: 450, 3354). The IR, 1H NMR, and MS of 2a closely resembled those of 1. On treatment with 3,5-dinitrobenzoyl chloride, 2a gave a bis-3,5-dinitrobenzoate (2c), which confirmed that the two oxygen atoms were present as hydroxy groups. Thus 2a

possessed an acyclic tetraprenyl structure with two hydroxymethyl groups, one of which was placed on C-1, in view of its splitting pattern (doublet) in the ¹H NMR spectrum. 2a was isomeric to 1 regarding the location of the extra hydroxyl group.

In order to determine the location of the hydroxyl group and the geometries of the double bonds in 2a, detailed decoupling experiments were attempted by use of Eu(DPM)₃ as a shift reagent. By comparison with the ¹HNMR spectrum of 1, all of the protons of 2a were assigned as listed in Table 1. On irradiation of the signal at δ 5.21 (C-14) in the presence of the shift reagent, both broad singlets at 1.65 and 1.69 (C-16 and C-20) were sharpened showing that these two methyl groups and the olefinic proton were arranged on the same double bond and mutually underwent an allylic long-range coupling. Therefore, the hydroxyl group was situated neither on C-16 nor C-20. On the other hand, as irradiation of the signal at 6.44 (C-2) resulted in sharpening of the signal at 2.02 (C-17) and the collapsing of the doublet at 5.48 (C-1) to a singlet, the hydroxyl group could not be placed on C-17. Consequently, it must be located on C-19.

The geometry of the Δ^2 -double bond was assigned as E by the $\Delta\delta$ values $\{\Delta\delta = \delta[\text{Eu}(\text{DPM})_3] - \delta(\text{CCl}_4)\}$ (see Table 1) for the methyl (C-17) over the methylene groups (C-4)[1] and by the chemical shift of the former [15]. The chemical shift of the olefinic proton (C-10) and the shift